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consists of almost all the

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difference between
chemical and physical
change. Describe the
difference between
accuracy and precision.
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matter, and give an
example of each

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Define relative atomic,
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masses, based on the ^{12}C scale and concept of isotopes. b) Explain mole in terms of the Avogadro ' s constant.
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2, and 3 1. In 1928, 3.3 g

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of a new element was isolated from 660 kg of the ore molybdenite. The percent by mass of this element in the ore was: a. 5 % b. 6.6 % c. 3.3 % d. 5.0×10^{-4} % e. 2.2 %

2. A quantitative observation
a. contains a number and a unit. b. does not contain a number.

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